

## TRIMETHOXYBENZENE DERIVATIVES FROM INDIGENOUS DRUGS\*

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Our earlier work on the isolation of active principle from the rhizomes of well known indigenous plant (1, 26) *Acorus calamus* Linn. (Uragandha, Bach) led us to investigate its volatile (4, 7, 8, 9, 18, 19, 20, 23, 24) which showed considerable pharmacological activity. Subsequent work on the fractionation (2) of the active substance from the volatile oil, by gas phase chromatography showed the presence of at least two components, (3, 6) which were isolated in pure state i.e.  $\alpha$ -asarone and  $\beta$ -asarone (Fig 1) which are trans-and cis-isomers, respectively, of 2, 4, 5-trimethoxy-1-propenyl benzene.

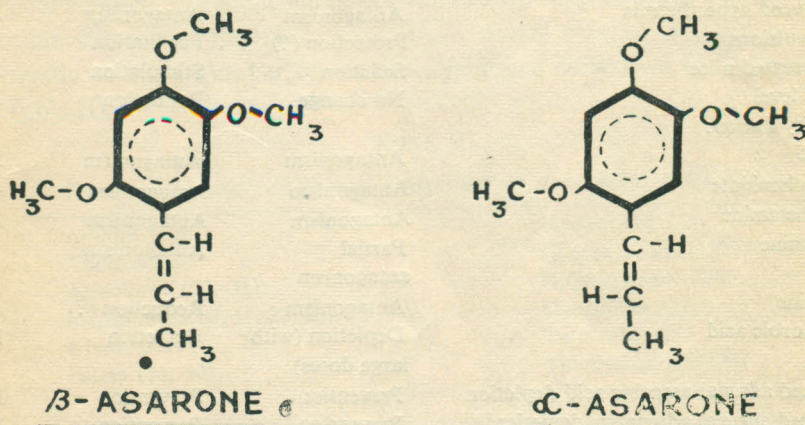


Fig 1

Asarone was found to possess very potent psychotropic properties (13, 15, 16). Asarone possessed a number of actions similar to reserpine and chlorpromazine (Table I). Besides the drug increased the number of shocks accepted by the animals in conflict neurosis (5).

But unlike reserpine it protected animals from electric shock (11) and metrazol induced convulsions and failed to cause release of 5-HT from the brain. The drug also prevented the depletion of adrenal ascorbic acid in rats subjected to cold stress (12).

These findings drew our attention to other naturally occurring drugs having a trimethoxybenzene function in their molecule. Of these at least, two i.e. reserpine and mescaline (Fig 2) are worth mentioning.

An important question that arises out of this study is that two fairly similar chemical structures i.e. asarone and mescaline have not only so different but quite opposite types of

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TABLE I

*Comparison of pharmacological actions of asarone with reserpine and chlorpromazine.*

Parameter	Asarone	Reserpine	Chlorpromazine
1. Spontaneous activity in rats, mice and monkeys	Reduction	Reduction	Reduction
2. Barbiturate hypnosis	Potentialiation	Potentialiation	Potentialiation
3. Behaviour of cats, monkeys	Taming	Taming	Taming
4. Fighting behaviour of mice	Suppression	Suppression	Suppression
5. Amphetamine toxicity in aggregated mice	Protection	Protection	Protection
6. Conditioned avoidance response	Specific blockade	Specific blockade	Specific blockade
7. Choice discrimination behaviour	Anxiety reduction	Anxiety reduction	Anxiety reduction
8. Rectal temperature	Hypothermia	Hypothermia	Hypothermia
9. LSD-25 induced hyperthermia	Antagonism	Antagonism	Antagonism
10. Electro-convulsions	Protection (?)	Facilitation	Facilitation (?)
11. Iproniazid-treated mice	Sedation	Stimulation	Sedation
12. Brain 5-HT level	No change	Reduction	No change
13. Hyperactivity due to:			
(a) LSD-25	Antagonism	Antagonism	Antagonism
(b) Methylphenidate	Antagonism	Antagonism	Antagonism
(c) d-Amphetamine	Antagonism	Antagonism	Antagonism
(d) Imipramine	Partial antagonism	Antagonism	—
(e) Mescaline	Antagonism	Reduction (?)	Antagonism
14. Adrenal ascorbic acid	Depletion (with large doses)	Depletion	Depletion
15. Stress-induced adrenal ascorbic acid depletion	Prevention	Prevention	Prevention
16. Stress-induced adrenal adrenaline depletion	Prevention	Prevention	Prevention
17. Stress-induced brain neurohormonal changes	Prevention	Prevention	Prevention
18. Tremorine-induced tremors.	Partial protection	Partial protection	Partial protection
19. $\alpha$ -Methyltyrosine treated animals:			
(a) Pentobarbitone hypnosis	Further potentialiation	Further potentialiation	Further potentialiation
(b) Conditioned avoidance response.	-do-	-do-	-do-
(c) Rectal temperature	-do-	-do-	-do-
20. Blood pressure	Hypotension	Hypotension	Hypotension
21. Smooth muscle contractions induced by:			
(a) Acetylcholine	Antagonism	Antagonism	Antagonism
(b) Histamine	Antagonism	—	Antagonism
(c) 5-hydroxytryptamine	Antagonism	Antagonism (?)	Antagonism



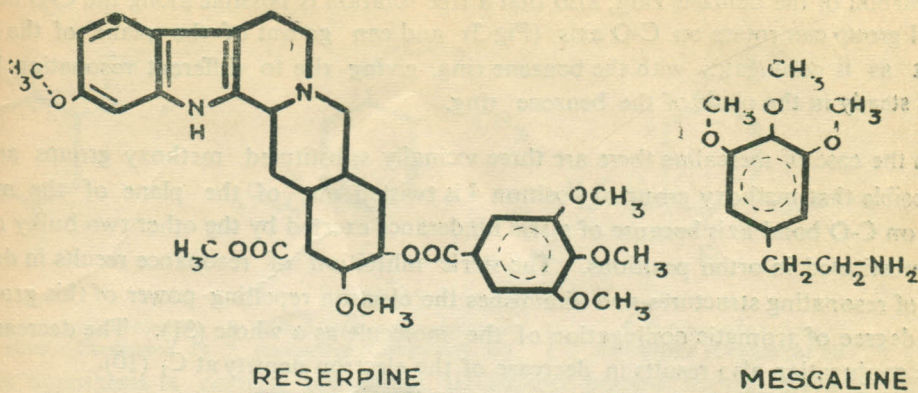


Fig 2

pharmacological actions. This intriguing phenomenon prompted us to explain the difference on the basis of molecular orbital theory.

To investigate this a number of trimethoxybenzene derivatives were synthesized in our laboratory (Table II).

TABLE II

*Trimethoxybenzene derivatives synthesised in the author's laboratory.*

Amides (30)	Hydantoins (28)
Anilides (30)	Hydrocarbons (17)
Azlactones (27, 31)	Nitro-styrenes (29)
Benzamides (10, 17, 22, 25, 32)	Oxazolines (27)
Cinnamides (17, 29)	Tetrazoles (27)
Esters (14, 30)	Thiohydantoins (28)

Pharmacological screening of these derivatives revealed that the position of the methoxy group in the benzene ring as well as unsaturation in the side chain played an important role in the manifestation of CNS depressant activity (17).

The recent electron diffraction studies and dipole moment data (21) have shown that a methoxy group, attached to the benzene ring is not straight like  $-O-CH_3$  but the methyl group is deviated at an angle of  $72^\circ$  from the axis of the C-O bond between the oxygen of methoxy group

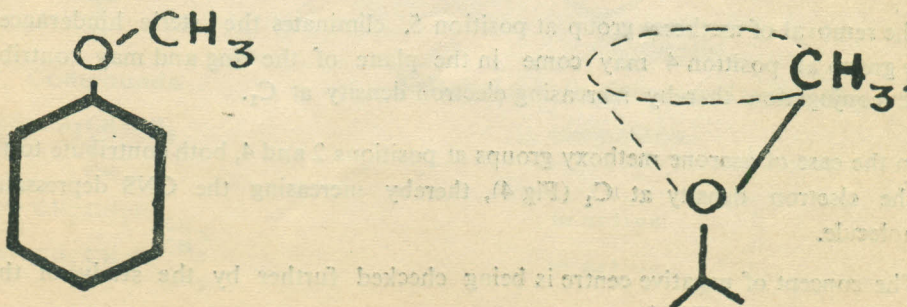


Fig 3



and the carbon of the benzene ring; also that a free rotation is possible along the C-O bond and so methyl group can rotate on C-O axis (Fig 3) and can go out of the plane of the benzene ring, but as it conjugates with the benzene ring, giving rise to different resonating hybrids, it is kept steady in the plane of the benzene ring.

In the case of mescaline there are three vicinally substituted methoxy groups and it is quite possible that methoxy group at position 4 is twisted out of the plane of the molecule, rotating on C-O bond axis because of steric hinderance exerted by the other two bulky methoxy groups substituted in ortho positions. The steric inhibition of resonance results in decreased number of resonating structures and diminishes the electron repelling power of this groups and also the degree of aromatic conjugation of the molecule as a whole (31). The decrease in the aromatic conjugation also results in decrease of the electron density at  $C_1$  (10).

The increase in the electron density at  $C_1$  is probably directly proportional to the CNS depressant potency. This view is further supported by the fact that removal of methoxy group at position 5 in mescaline results in 50% loss of the stimulant activity (33).

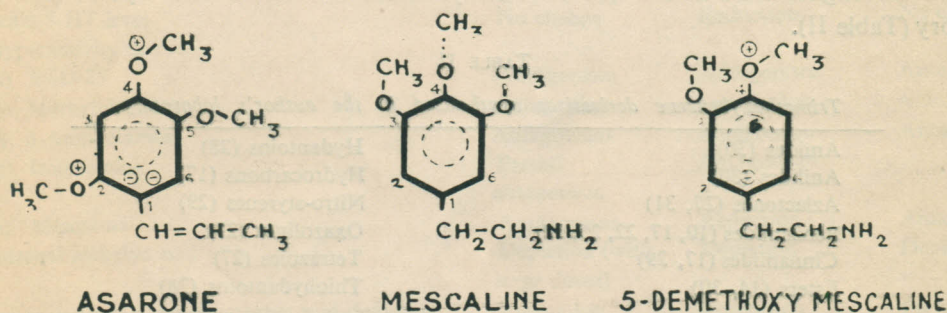


Fig 4

(In Fig 4 two -ve signs only represent comparative increase in the electron density at  $C_1$ , they are not true negative charges and the total value of these charges is only a fraction of true electron charge).

The removal of methoxy group at position 5, eliminates the steric hinderance and then methoxy group at position 4 may come in the plane of the ring and may contribute to the aromatic conjugation thereby increasing electron density at  $C_1$ .

In the case of asarone methoxy groups at positions 2 and 4, both contribute to the increase in the electron density at  $C_1$  (Fig 4), thereby increasing the CNS depressant activity of the molecule.

The concept of negative centre is being checked further by the study of the recently synthesised compounds (Fig 5).



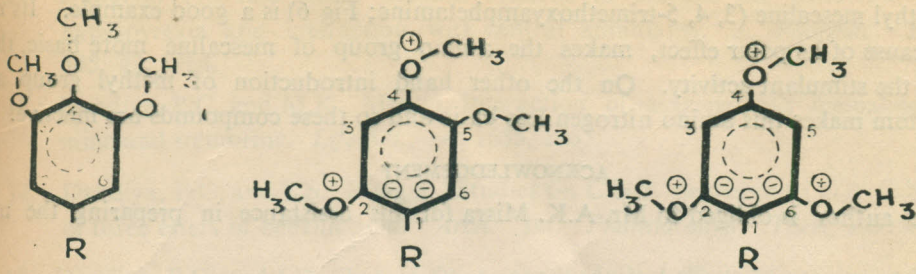


Fig 5

3, 4, 5-trimethoxybenzene derivatives (Compound I) | 2, 4, 5-trimethoxybenzene derivatives (Compound II) | 2, 4, 6-trimethoxybenzene derivatives (Compound III)

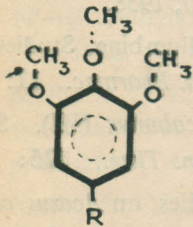
If this hypothesis is correct, compound III should possess highest CNS depressant activity, as in this compound all the three methoxy groups contribute to the increase in the electron density at C<sub>1</sub>. (This is represented by three negative signs in the figure).

There are certain factors other than conjugation like hyperconjugation due to the double bond and -I effect due to the methyl group in the side chain (in asarone) which might also play some role. But these effects would not be very significant.

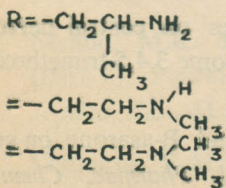
Apart from these considerations, the nature of side chain is also responsible for CNS depressant activity. It was observed in this laboratory (17) that increase in the length of side chain reduces CNS depressant activity whereas branching in the side chain increases CNS depressant activity. This can be explained on the basis of inductomeric effect.

With the increase in the length of side chain the inductomeric effect of certain specific groups goes on decreasing and it cannot conjugate with the trimethoxybenzene nucleus, whereas, branching of the side chain increases the polarity of the molecule, thereby contributing to the inductomeric effect of these groups present in the side chain.

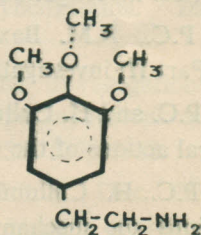
3,4,5-Trimethoxybenzene derivatives



Compounds



Mescaline



Biological activity as compared to mescaline

more active

in active

in active

Fig 6



Methyl mescaline (3, 4, 5-trimethoxyamphetamine; Fig 6) is a good example. Its methyl group, because of its polar effect, makes the amino group of mescaline more basic, thereby increasing the stimulant activity. On the other hand introduction of methyl group at the nitrogen atom makes this amino nitrogen less basic and so these compounds are inactive.

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